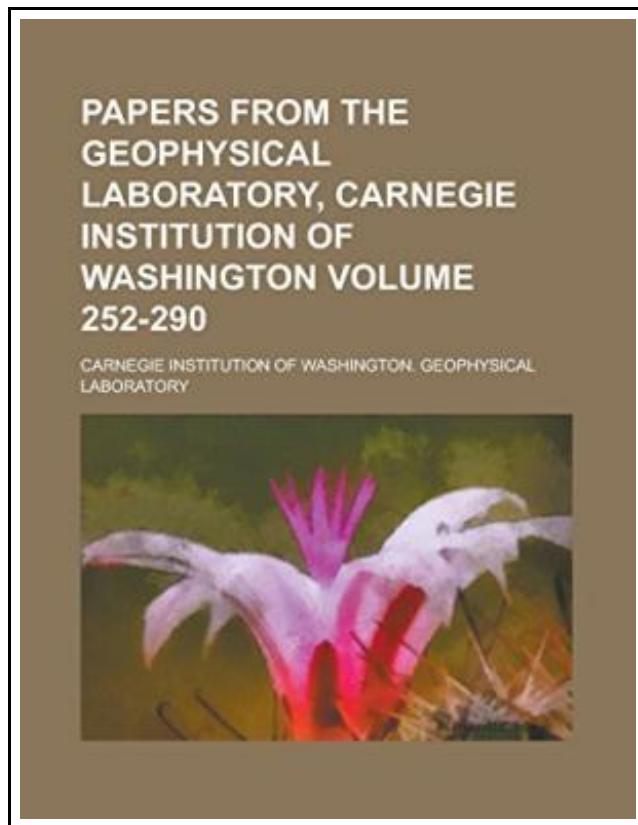


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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1918 edition. Excerpt: .binary system has been drawn on this basis; it is the broken curve in Fig. 6. The eutectic K₂SiO₄-K₂Si₂O₅ is at about 7750 and a SiO₂/K₂O ratio of 1.30, the K₂SiO₄-SiO₂ eutectic at 525 1 Niggli (this Journal, 35, 1710 (1913)) gives 1015 for the melting point of K₂SiO₄ and a SiO₂/K₂O ratio of 4.0. The broken curve representing the equilibrium of melts containing small amounts of K₂SiO₄ with SiO₂ in one of its forms as solid phase is wholly an extrapolation. Its end points at the K₂SiO₄-a-quartz eutectic, and the melting temperature of pure silica are known, however, and the true curve cannot deviate much from its assumed course. It also must show breaks corresponding to the change K₂SiO₄ 1.25 15 2.0 3.0 4.0 6.0. Stf2 RATIO S.02/K20 Fig. 6. Diagram showing the variation of melting point with SiO₂/K₂O ratio on the one atmosphere isobar (cf. Fig. 4). from -quartz to /3-quartz at 575, and from /3-quartz to tridymite at 8700; the inversion tridymite-cristobalite at 14700 has also been considered, though this point does not lie inside the diagram as given. C. The Binary Systems H₂O-K₂O, EUO-K₂Si₂O₅, K₂SiO₄-SiO₂, and H₂O-SiO₂.--The ternary system H₂O-K₂SiO₄-SiO₂ is limited by the binary systems K₂SiO₄-SiO₂, H₂O-SiO₂, and H₂O-K₂SiO₄. The first of these has already been discussed; its components are sufficiently non and is practically vertical. A second is the dissociation pressure curve of the higher hydrate. The course of these two curves is well known....

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