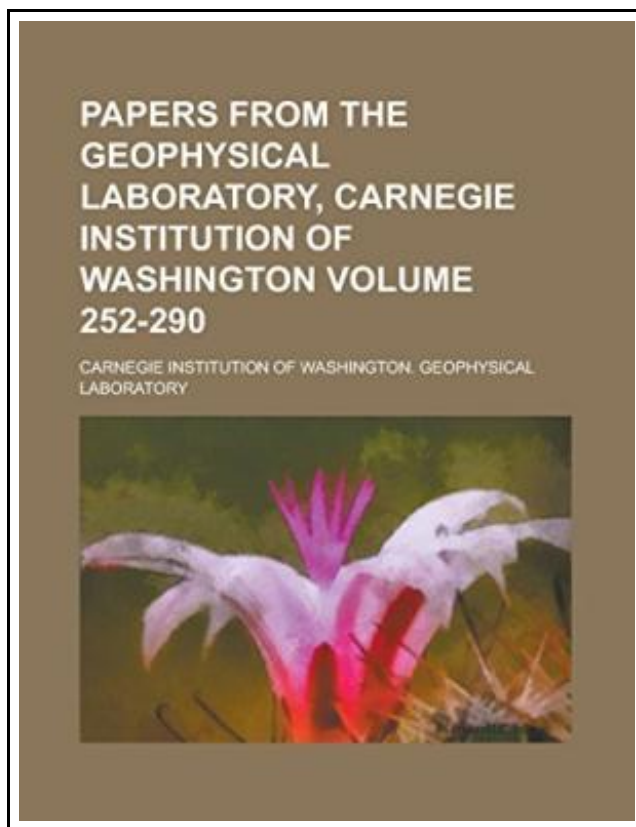


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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1918 edition. Excerpt: .binary system has been drawn on this basis; it is the broken curve in Fig. 6. The eutectic K<sub>2</sub>SiO<sub>3</sub>-K<sub>2</sub>SiO<sub>4</sub> is at about 7750 and a SiO<sub>2</sub>/K<sub>2</sub>O ratio of 1.30, the K<sub>2</sub>SiO<sub>3</sub>-SiO<sub>2</sub> eutectic at 525 1 Niggli (this Journal, 35, 1710 (1913)) gives 1015 for the melting point of K<sub>2</sub>SiO<sub>3</sub> and a SiO<sub>2</sub>/K<sub>2</sub>O ratio of 4.0. The broken curve representing the equilibrium of melts containing small amounts of K<sub>2</sub>SiO<sub>4</sub> with SiO<sub>2</sub> in one of its forms as solid phase is wholly an extrapolation. Its end points at the K<sub>2</sub>SiO<sub>3</sub>-a-quartz eutectic, and the melting temperature of pure silica are known, however, and the true curve cannot deviate much from its assumed course. It also must show breaks corresponding to the change K<sub>2</sub>SiO<sub>3</sub> 1.25 15 2.0 3.0 4.0 6.0. Stf2 RATIO SiO<sub>2</sub>/K<sub>2</sub>O Fig. 6. Diagram showing the variation of melting point with SiO<sub>2</sub>/K<sub>2</sub>O ratio on the one atmosphere isobar (cf. Fig. 4). from -quartz to /3-quartz at 575, and from /3-quartz to tridymite at 8700; the inversion tridymite-cristobalite at 14700 has also been considered, though this point does not lie inside the diagram as given. C. The Binary Systems H<sub>2</sub>O-K<sub>2</sub>SiO<sub>3</sub>, EUO-K<sub>2</sub>SiO<sub>4</sub>, K<sub>2</sub>SiO<sub>3</sub>-SiO<sub>2</sub>, and H<sub>2</sub>O-SiO<sub>2</sub>.--The ternary system H<sub>2</sub>O-K<sub>2</sub>SiO<sub>3</sub>-SiO<sub>2</sub> is limited by the binary systems K<sub>2</sub>SiO<sub>3</sub>-SiO<sub>2</sub>, H<sub>2</sub>O-SiO<sub>2</sub>, and H<sub>2</sub>O-K<sub>2</sub>SiO<sub>3</sub>. The first of these has already been discussed; its components are sufficiently non and is practically vertical. A second is the dissociation pressure curve of the higher hydrate. The course of these two curves is well known....



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